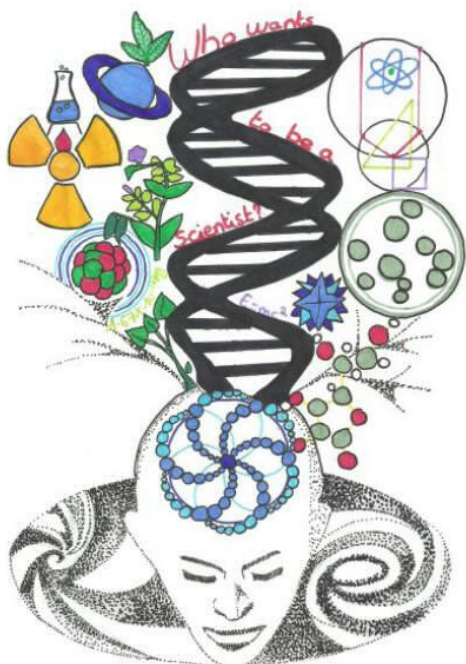
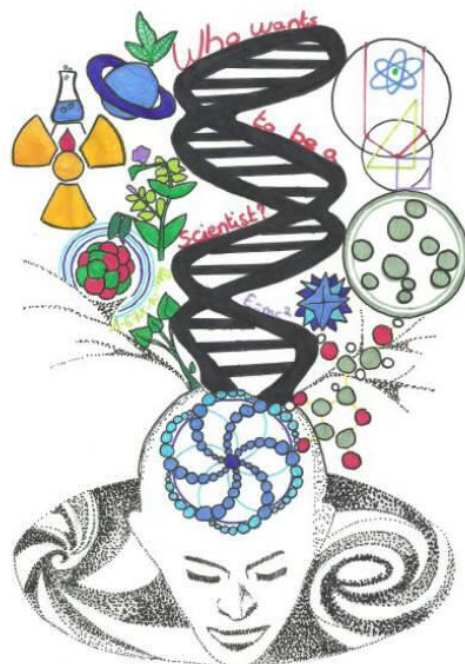


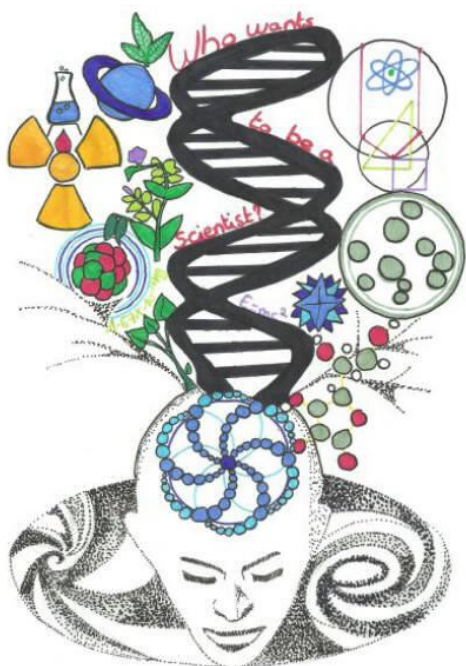
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PERIODIC ELEMENTS



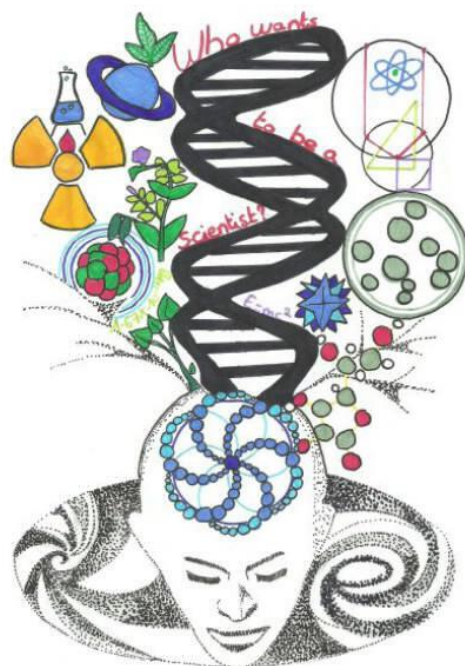
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PERIODIC ELEMENTS



## PERIODIC ELEMENTS

This element is present in milk. It has atomic number 20, belongs to the group of alkaline earth metal soft metal. It is the fifth element in order of abundance in the earth's crust. It reacts with water.

Calcium

## PERIODIC ELEMENTS

This element was discovered by the Swedish chemist Axel Fredrik Cronstedt. It is a hard, corrosion resistant metal. It is alloyed with other metals to improve their strength and resistance to corrosion. At room temperature it is solid and it's white. It's a natural element and its number of protons is 28.

Nickel

## PERIODIC ELEMENTS

It's a soft metal, dense, ductile and malleable. It's white but it becomes dark gray when it comes in contact with air. It's both artificial and natural. At room temperature it's solid. In its nucleus there are 82 protons. It was discovered by Egyptians in 1550 a.C..

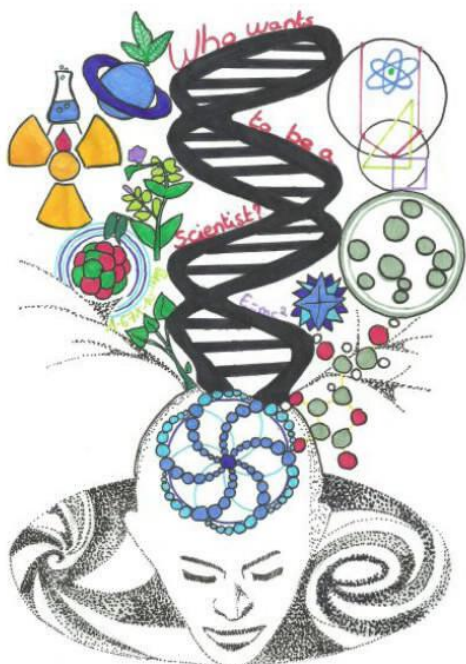
Lead

## PERIODIC ELEMENTS

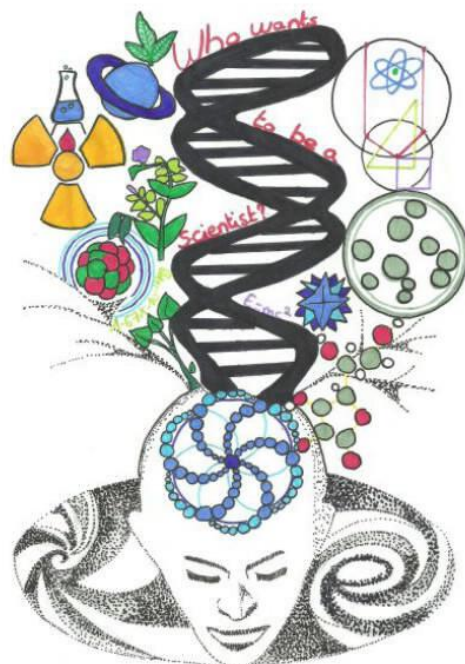
It is a hard metal, shiny, steel gray; it can be easily polished, very resistant to corrosion. It is solid at room temperature. It is natural and not artificial. Atomic number 24. It was discovered by Johann Lehmann in 1761.

Chromium

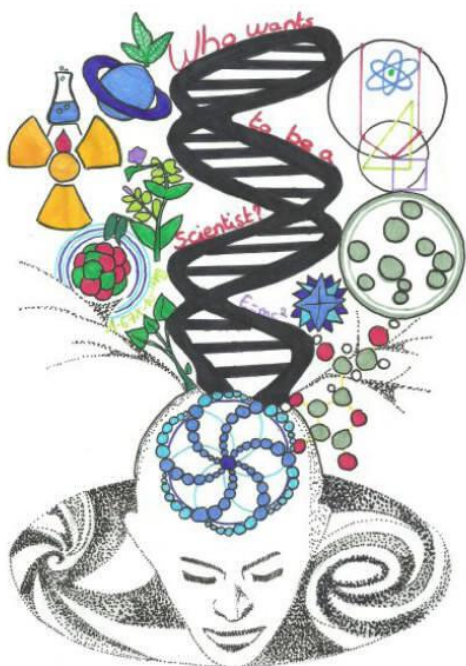
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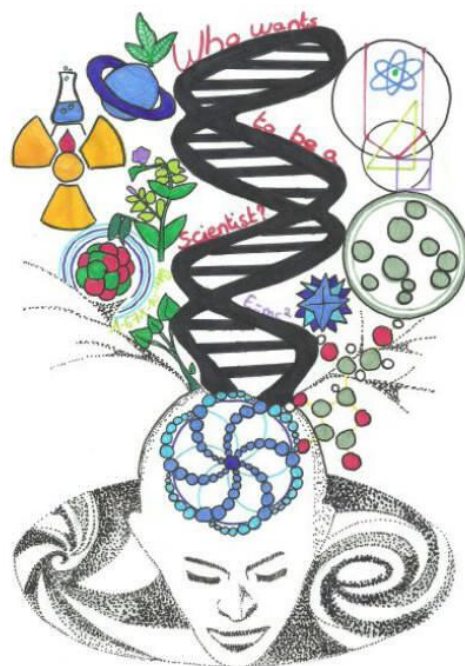
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## PERIODIC ELEMENTS

The element used in nuclear power stations to produce energy.

It has atomic number 92. It is a silvery-white metal in the actinide series.

It has the second highest atomic weight in the primordially occurring elements. In nature is found in different isotopes, and decays slowly by emitting alpha particle.

It was discovered by Martin Heinrich Klaproth in 1789.

Uranium